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Any polyatomic ion with the suffix "-ate" uses the suffix "-ite" as an acid. So, HNO₃ will be nitric acid. When you have a polyatomic ion with one more oxygen than the "-ate" ion, then your acid will have the prefix "per-" and the suffix "-ic." For example, the chlorate ion is ClO₃⁻. Therefore, HClO₄ is called perchloric acid.

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