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A)8, 2, and 3 B)7, 1, and 2 C)2, 7, and 4 D)2, 4, and 6 E)1, 5, and 7 6) Lattice energy is \_\_\_\_\_. A)the sum of electron affinities of the components in an ionic solid B)the energy required to convert a mole of ionic solid into its constituent ions in the gas phase C)the sum of ionization energies of the components in an ionic solid

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Flow Chart for selection of test for gas

8. Test For Anions ANIONS TEST 1

Carbonate ion,  $\text{CO}_3^{2-}$ -- Pour 3 cm<sup>3</sup> of aqueous carbonate solution into a test tube - Add 2 cm<sup>3</sup> hydrochloric acid, HCl - Bubble the gas produced through limewater -Lime water turned milky.

2 Chloride ion,  $\text{Cl}^-$ -- Pour about 2 cm<sup>3</sup> of aqueous chloride solution into test tube

## *Chapter 8: Salts - Pasti Skor*

Material for test #1 . Chapter 1  
Chapter 2 Chapter 3 Chapter 5 .  
Material for test #2 . Chapter 4  
Chapter 6, part a Chapter 6, part b  
Chapter 6, part c Representing  
chemical reactions . Material for test  
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## *Organic Chemistry I*

Chem Chapter 8 Test. Ashley J. • 30. cards. If metal X is lower than metal Y in the activity series, then metal X. loses electrons less readily than does metal Y. A chemical equation is balanced when the. Same number of each kind of atom appears in the reactants and in the products.

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Chapter 8 Review Multiple Choice

Identify the choice that best completes the statement or answers the question.

\_\_\_ A\_\_\_ 1. In a chemical reaction a. the mass of the reactants equals the mass of the products. b. the mass of the products is greater than the mass of reactants. c. the number of atoms in...

<https://studyres.com/doc/2433046/honors-chemistry-chapter-8-test-review-answer-key>.

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