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ANSWER Answer the

following questions in the

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Review Physical
Characteristics
Of Gases

Pressure can be represented by the following equation:

pressure increase a. For a constant area, as the force increases the pressure will. decrease b.

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ANSWER Answer the

following questions in the space provided. 1.

Pressure can be represented by the following equation:

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constant force, as the area increases the pressure will

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ANSWER Answer the following questions in the space provided. 1.

Identify whether the descriptions below describe an ideal gas or a real gas. a. Gas particles move in straight lines until they collide with other particles or the walls of their container.

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~~Physical Characteristics~~

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10-1 The Kinetic-
Molecular Theory of
Matter. I. The Kinetic-
Molecular Theory of
Gases A. Ideal Gas 1. An
imaginary gas that
perfectly fits all the
assumptions of the
kinetic- molecular theory

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B. Five Assumptions of the Kinetic-Molecular Theory

1. Gases consist of large numbers of tiny particles that are far apart relative to their size

2.

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are not being met
sufficiently with current
toys. Will help balance
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1. Chemistry S. Martinez
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KMT – based on idea
that particles of matter
are always in motion. 2.

3. 1. Gases consist of
large numbers of tiny
particles that are far apart
relative to their size. 2.

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. decrease b. For a
constant ...

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